

Application of Statistical Mechanics

Imperfect gases

Lecture 3: Second Virial Coefficient

Lecture notes & slides: <http://dullens.chem.ox.ac.uk/main/teaching.html>

Recap of 2nd lecture

- Classical Statistical Mechanics
 - 6N-dimensional phase space

In 3D this means that 6 coordinates must be specified for each particle

$$(p_x, p_y, p_z, x, y, z) = (p, \tau)$$

Recap of 2nd lecture

- Classical Statistical Mechanics
 - 6N-dimensional phase space
 - Classical molecular partition function

$$q = \sum_i e^{-\beta \varepsilon_i}$$



$$q_{class} = \frac{1}{h^3} \int \dots \int e^{-\beta H(p, \tau)} dp d\tau$$

Classical energy $H(p, \tau)$

Recap of 2nd lecture

- Classical Statistical Mechanics
 - 6N-dimensional phase space
 - Classical molecular partition function
 - Classical canonical partition function

$$Q_{class} = \frac{1}{N!h^{3N}} \int \cdots \int e^{-\beta H_N(p,\tau)} dp d\tau$$

$H_N(p,\tau)$: Classical N-body Hamiltonian for **interacting** particles

Recap of 2nd lecture

- Classical Statistical Mechanics
 - 6N-dimensional phase space
 - Classical molecular partition function
 - Classical canonical partition function
 - Configuration integral (including interactions!)

$$Q_{class} = \frac{1}{N!} \left(\frac{2\pi m k_B T}{h^2} \right)^{\frac{3N}{2}} Z_N$$

$$Z_N = \int_V e^{-\beta U(x_1, \dots, z_N)} dx_1 \cdots dz_N$$

Fundamental equations in the study of monatomic, classical imperfect gases and liquids

Content of the course

- Canonical Ensemble (recap)
- Classical Statistical Mechanics
- Second Virial Coefficient B_2
- B_2 for imperfect gases

Imperfect gases: Virial expansion

$$p = k_B T (\rho + B_2 \rho^2 + B_3 \rho^3 + \dots)$$

B_2 : second virial coefficient

(1st year states of matter)

$$B_2(T) = -2\pi \int_0^{\infty} (e^{-\beta U(r_{ij})} - 1) r^2 dr$$

Last two lectures: derive this expression for B_2

Strategy to obtain B_2

→
$$Z_N = \int_V e^{-\beta U(x_1, \dots, z_N)} dx_1 \cdots dz_N$$

$$Q_{class} = \frac{1}{N!} \left(\frac{2\pi m k_B T}{h^2} \right)^{\frac{3N}{2}} Z_N$$

$$p = k_B T \left(\frac{\partial \ln Q}{\partial V} \right)_{T, N}$$

$$p = k_B T (\rho + B_2 \rho^2 + B_3 \rho^3 + \dots)$$

Pairwise additivity

- Potential energy between pair only depends on distance

$$U_{ij} = U(r_{ij})$$

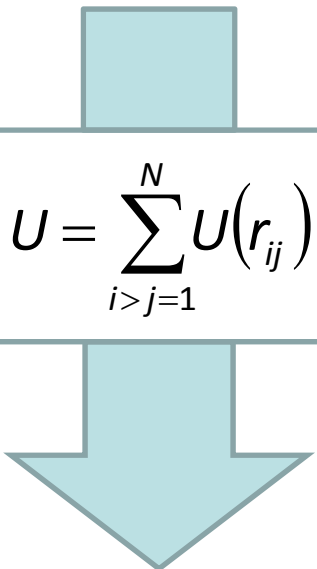
- Total potential energy = sum of potential energy contributions between each pair of molecules

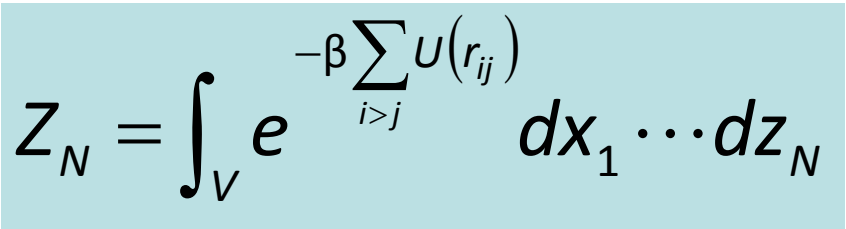
$$U_{tot} = \frac{1}{2} \sum_{i \neq j} U(r_{ij})$$

$$U_{tot} = \sum_{i > j} U(r_{ij})$$

Classical Configuration Integral Z_N

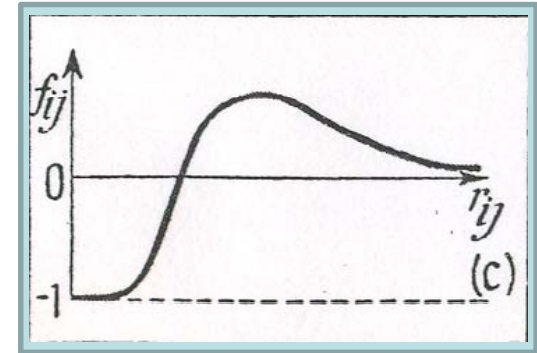
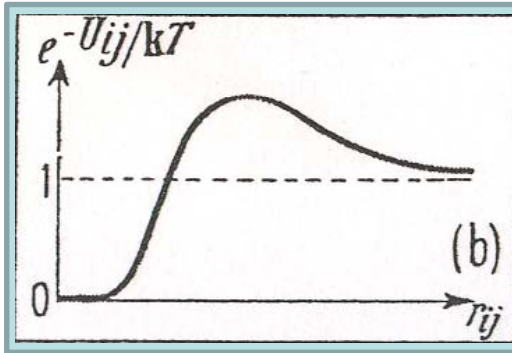
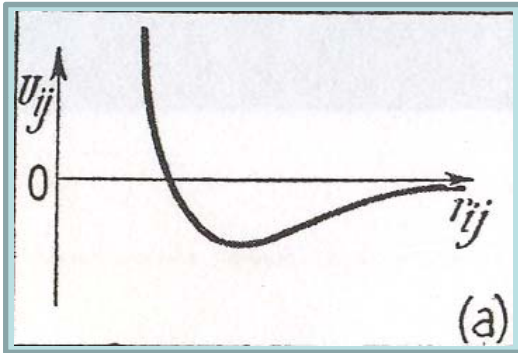
$$Z_N = \int_V e^{-\beta U(x_1, \dots, z_N)} dx_1 \cdots dz_N$$


$$U = \sum_{i>j=1}^N U(r_{ij})$$


$$Z_N = \int_V e^{-\beta \sum_{i>j} U(r_{ij})} dx_1 \cdots dz_N$$

Cracking $Z_N \dots$: Mayer f -function $f(r_{ij})$

$$f(r_{ij}) \equiv e^{-\beta U(r_{ij})} - 1$$



f_{ij} vanishes for large distances, i.e., when two particles are far apart

$$Z_N = \int \cdots \int \prod_{i>j} (1 + f_{ij}) d\tau_1 \cdots d\tau_N$$

Expanding the product ...

Only writing down terms that involve one or less Mayer function f_{ij}

Example: $N=3$

$$\begin{aligned}\prod_{i>j}^{N=3} (1 + f_{ij}) &= (1 + f_{21})(1 + f_{31})(1 + f_{32}) \\ &= (1 + f_{21} + f_{31} + f_{21}f_{31})(1 + f_{32}) \\ &= 1 + \underbrace{f_{21} + f_{31} + f_{32}}_{\text{pair-terms}} + \underbrace{f_{21}f_{31} + f_{21}f_{32} + f_{31}f_{32} + f_{21}f_{31}f_{32}}_{\text{higher order terms}}\end{aligned}$$

General:

$$\prod_{i>j}^N (1 + f_{ij}) = 1 + \sum_{i>j} f_{ij} + \dots$$

Z_N in terms of Mayer f -functions

$$Z_N = \underbrace{\int \cdots \int d\tau_1 \cdots d\tau_N}_{\text{perfect gas}} + \underbrace{\int \cdots \int \sum_{i>j} f_{ij} d\tau_1 \cdots d\tau_N}_{\text{corrections to perfect gas}} + \dots$$

perfect gas

corrections to perfect gas



V^N



continue ...

Z_N continued ...

$$Z_N = V^N + \underbrace{\int \cdots \int \sum_{i>j} f_{ij} d\tau_1 \cdots d\tau_N}_{\text{... each pair gives the same contribution ...}} + \dots$$

... each pair gives the same contribution ...

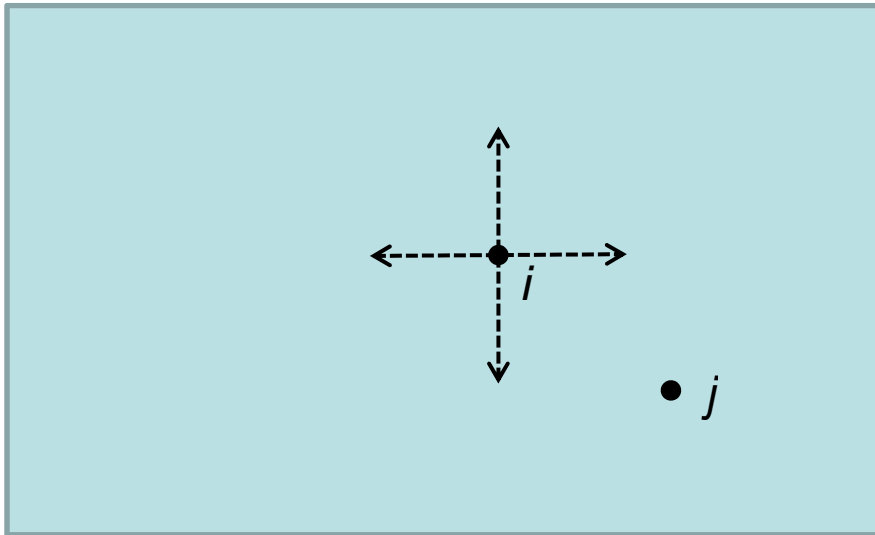
... $N(N-1)/2$ pairs ...



$$Z_N = V^N + \frac{N(N-1)}{2} \underbrace{\int \cdots \int f_{ij} d\tau_1 \cdots d\tau_N}_{\text{crack this (1)}}$$

Crack (1) continued ...

$$\int \cdots \int f_{ij} d\tau_1 \cdots d\tau_N = V^{N-2} \underbrace{\int \int f_{ij} d\tau_i d\tau_j}_{\text{crack this (2)}}$$



- choose position of particle j
- integrate over all positions of i
- integral same for all positions of j
- integral independent of position of j

So ...

$$Z_N = V^N + \frac{N(N-1)}{2} \underbrace{\int \cdots \int f_{ij} d\tau_1 \cdots d\tau_N}$$

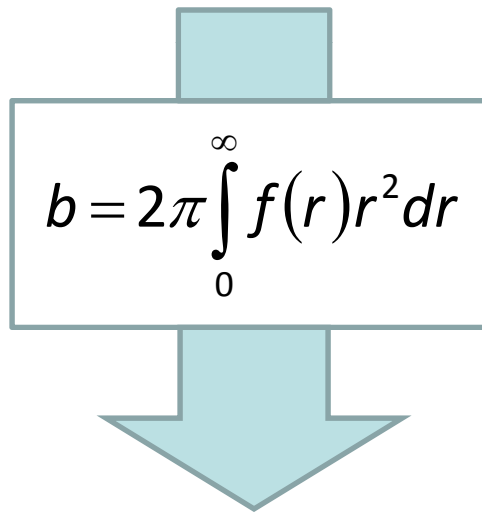
$$V^{N-1} \int_{-\infty}^{\infty} 4\pi f(r) r^2 dr$$



$$Z_N = V^N + \frac{N(N-1)}{2} V^{N-1} \int 4\pi f(r) r^2 dr + \dots$$

Short-hand notation for integral

$$Z_N = V^N + \frac{N(N-1)}{2} V^{N-1} \int 4\pi f(r) r^2 dr + \dots$$


$$b = 2\pi \int_0^{\infty} f(r) r^2 dr$$

$$Z_N = V^N + N(N-1) V^{N-1} b + \dots$$

Strategy to obtain B_2

\checkmark

$$Z_N = \int_V e^{-\beta U(x_1, \dots, z_N)} dx_1 \cdots dz_N$$

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lecture 4